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1900s scientists discovered that various  
isotopes will undergo nuclear decay.  
During this process the unstable nucleus  
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scientists studied this radiation they discovered three types of particles: alpha, beta, and gamma.

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To describe a nuclear reaction, we use an  
equation that identifies the nuclides  
involved in the reaction, their mass



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numbers and atomic numbers, and the other particles involved in the reaction.

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nuclear equations that describe the  
following processes. 1. Uranium-235

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undergoes an alpha decay to produce thorium-231. 2. Lanthanum -144 becomes cerium-144 when it undergoes a beta decay. 3. Neptunium-233 is formed when americium-237 undergoes a nuclear decay process. 4.

Nuclear Radiation Worksheets - Lesson

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Nuclear chemistry is the subfield of chemistry dealing with radioactivity, nuclear processes, such as nuclear transmutation, and nuclear properties. It is the chemistry of radioactive elements such as the actinides, radium and radon together with the chemistry associated

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with equipment (such as nuclear reactors) which are designed to perform nuclear processes.

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