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6. Determine the molarity of a solution prepared by dissolving 11.85 g
of KMnO_4 in enough water to make 750.0 mL of solution. (5 points) 7. A
0.1025g sample of copper metal is dissolved in 35.0 mL of concentrated
nitric acid forming Cu^{2+} ions in solution. The solution is then
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 Cu^{2+} .*

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