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Section Focus Transparency 20 and Master Teaching Transparency 18 and Master Study Guide for Content Mastery, pp. 31-32 TCR Multimedia Resources Chemistry Interactive CD-ROM, Section 6.1 Exploration, Video, and Demonstration MindJogger Videoquizzes, Ch. 6 Guided Reading Audio Program, Section 6.1 Cosmic Chemistry Videodisc, Disc 3, Side 6 ...

UNIT 3 RESOURCES Surface Processes on Earth

TEACHING TRANSPARENCY 18 Chemistry: Matter and Change Teaching Transparency Worksheet 2 Use with Chapter 6, Section 6.1 The Periodic Table 1. How many elements are listed in the periodic table? _____ 2. What is the atomic number of selenium? ...

The Periodic Table and Periodic Law - Glencoe

86 Chemistry: Matter and Change • Chapter 4 Teaching Transparency Masters Electrically charged plates 2

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Name Date Class TEACHING TRANSPARENCY

Name Date Class Calculating Numbers of Electrons Use with Chapter 7, Section 7.1 and Predicting Ionic
Change 1. What happens to a neutral atom if it loses one electron?

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Teaching Transparency 22 - Formation of Ions 1. calcium and oxygen 2. Yes; each contains equal numbers
of protons and electrons. 3. ionization energy 4. positive; cation 5. negative; anion 6. No; it is not a
stable octet of electrons. 7. No; it is not a stable octet of electrons. 8. Yes; it is a stable octet of
electrons. 9.

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10 Chemistry: Matter and Change • Chapter 5 Teaching Transparency Masters 4s 3p 3s 2p 2s 7p 5f 6p 4f 5p
5s 3d 7s 5d 6s 4d 4p 7p 6p 6d 5f 4f 5d 4d 3d 4p 3p 2p 6d 7s 6s 5s 4s 3s 2s 1s 5p 1s Increasing Energy
Orbital filling sequence Orbital Filling Sequence and Energy Levels TEACHING TRANSPARENCY MASTER Use
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TEACHING TRANSPARENCY

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UNIT 1 RESOURCES Earth Science - Dearborn Public Schools

TEACHING TRANSPARENCY WORKSHEET The Activity Series Use with Chapter 9, Section 9.2 1. For each of the following pairs of elements, underline the one that would replace the other element in a compound. a. calcium, tin e. iron, copper b. bromine, fluorine f. iodine, chlorine c. aluminum, potassium g. silver, lead d. zinc, sodium 2.

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Name Formation of Ions 1. What are the names of the two elements shown? Date Class 22 Use with Chapter 7, Section 7.1 2. Are the elements shown on the left sides of the two equations neutral?

TEACHING TRANSPARENCY MASTER 17 Orbital Filling Sequence ...

Teaching Transparency Transparency Master 18 Earth Science: Geology, the Environment, and the Universe 11 Stage 1. Weathering breaks solid bedrock into smaller pieces, forming the C horizon. Stage 2. Weathering of rock pieces continues. Organisms living in the weathered material die and decay, forming an organic-rich top layer, or A horizon.

TEACHING TRANSPARENCY MASTER 11 Cathode Ray Experiments ...

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Answer Teaching Transparency Worksheet 36

Teaching Transparency Worksheets Chemistry: Matter and Change • Chapter 6 35 1. What are the four sections, or blocks, of the periodic table? 2. What does each block represent? 3. What do elements in the s-block have in common? 4. What is the valence electron configuration of each element in group 1? 5.

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TEACHING TRANSPARENCY WORKSHEET The Activity Series

TEACHING TRANSPARENCY WORKSHEET Atomic Orbitals 1. What is the shape of an s orbital? Class 16 Use with Chapter 5, Section 5.2 2. What is the relationship between the size of an s orbital and the principal energy level in which it is found? 3. What is the shape of a p orbital? How many p orbitals are there in a ublevel? 4.

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