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Transparency. The answers can be found in Chapter 13.1 of the respective book.

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CHAPTER 12 Answer

Keys Teaching

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Global Wind Systems

1. the northeast trade winds in the northern

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hemisphere and the southeast trade winds in the southern hemisphere 2. Air rises at the equator and then flows northward. At about 30 degrees north latitude, the air sinks.

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MASTER 42 Solubilit

y-Temperature ...

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Chapter 3

matterproperties and

changes, Order of

operations pemdas

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UNIT 3 RESOURCES

Surface Processes

on Earth

8 Chemistry: Matter
and Change •

Chapter 5 Teaching

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orbital z x y 2s
z x px py pz p orbitals
d orbitals y z x y z x y
z x y dxy dxz dyz
dx²-y² dz² x y y x z x
y z z z x y Atomic
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with Chapter 2,

Visualizing the Use

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Section 2.4

Conservation of Mass

Section 3.2 1.

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Transparency 18-2

Connecting Math to

Physics FAST

FILE Chapters 16–20

Resources, Chapter

18 Transparency 18-3

Master, p. 93 ...

Analysis Answers will

vary. Sample

answers: The

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Key Help
Section/Objectives
Standards Lab and
Demo Planning

2. Are the elements shown on the left sides of the two equations neutral? How can you tell? 3. What is the name for the energy needed to remove electrons from an atom, such as the ... Teaching

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Worksheets
Chemistry: Matter and
Change Chapter 7 ...

Earth/Space Life

Cycle of a

Thunderstorm

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Objectives • Explain
the sequence of steps
used in solving
stoichiometric

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problems. • Use the steps to solve stoichiometric problems. Lesson Resources Section Focus Transparency 44 and Master Math Skills Transparencies 16–17 and Masters Teaching Transparency 38 and Master ChemLab and MiniLab Worksheets, pp. 45–48 TCR

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11 1. For each of the
following pairs of
elements, underline
the one that would

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replace the other
element in a
compound. a.
calcium, tin e. iron,
copper b. bromine,
fluorine f. iodine,
chlorine c. aluminum,
potassium g. silver,
lead d. zinc, sodium
2.

LESSON PLAN 12 -
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Key 10 Earth Science:

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Environment, and the

Universe 33 Mohs

Hardness Scale

MASTER 10

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Use with Chapter 4

Section 4.2 Mohs

Hardness Scale

Hardness of

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Hardness Common

Objects Talc 1

(softest) Gypsum 2

fingernail (2.5) Calcite

3 piece of copper

(3.5) Fluorite 4 iron

nail (4.5)

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WORKSHEET The
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Activity Series Use
with Chapter 9,
Section 9.2 1. For
each of the following
pairs of elements,
underline the one that
would replace the
other element in a
compound. a.

calcium, tin e. iron,

copper b. bromine,

fluorine f. iodine,

chlorine c. aluminum,

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potassium g. silver,
lead d. zinc, sodium
2.

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Explain your answer.

7. How many degrees of longitude separate Las Vegas from the International Date Line, which is at 180° longitude? 8

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Worksheet 4 Earth
Science: Geology, the
Environment, and the
Universe Teaching

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WORKSHEET 4

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Use with Chapter 2

Mapping Latitude and

Section 2.1 Longitude

UNIT 1 RESOURCES

Earth Science

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CHAPTER 12

Answer Key -

CHAPTER 12

Answer Keys

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MASTER 16 Math

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Masters Chemistry:

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Chapter 11 31

Master 2 Answer
Key Hp
number of moles of
known substance C H

C H C H O O O CO

CO CO H O H O H O

number of moles of

unknown substance O

CO H O C H CO H O

C H O H O C H O CO

use mole ratio 5 mol

O 2 1 mol C 3 H 8 3

mol CO 2 1 mol C 3 H

8 4 mol H 2 O 1 mol

...

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**Use with Chapter 15
Sonar - Science with
Mrs. Lambert**

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WORKSHEET Ionic
Bonds Use with
Chapter 7, Section
7.2 1. How many
valence electrons
does a neutral
magnesium (Mg)
atom have? 2. What

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is the charge on a
magnesium ion?

Master 2 Answer
MyUp
What does

magnesium have to
do to form such an
ion, and why does it
tend to do so? 3. How
many valence
electrons does a
single neutral chlorine
(Cl atom have? 4.

Use with Chapter 4

Crystal Systems -

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Deer Valley Unified

Master 2 Answer

...

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Read carefully the entire lab and then answer the following questions. Your

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begin. 1. ... Teaching

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16Earth Science:

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Universe7 Hydrolysis

2KAlSi₃O₈ 2(H

water carbon HCO₃)

H₂O Al₂Si₂O₅

(OH)₄ 2K H₂HCO₃

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WORKSHEET 31

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Geology, the

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Measurements in

Science MASTER 2

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Measurements in

Science Use with

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Mapping Latitude

and Section 2.1

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36 Chemistry: Matter
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Chapter 14 Teaching
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Masters 0 10 20 30

40 50 60 70 80 90

100 Solubility (g

solute/100 g H₂O)

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90 100 80 70 60 50

40 30 20 10 0

Temperature (oC)

Solubilities as a

Function of

Temperature NaCl

KClO₃ KCl CaCl₂

Ce₂(SO₄)₃ Solubility

–Temperature Graphs

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