

## Sweet Sixteen Chemistry Compound Tournament Answers

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### Sweet 16 Chemistry Compound Tournament - Flinn

tournament. This clever activity combines the ever-popular March Madness basketball pool with a review of chemical formulas, the structure and charges of ions, and molar mass calculations. The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important ...

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Sweet 16 Chemistry Ion Tournament First Round Second Round Semifinals Finals Winner NiCl2 Ni(II)2+ Fe(II) S Zn CO3 Na CrO4 Mg PO4 Ag(1) SO4 Ba NO3 Al OH Rules 1. First Round ?Add Charges 2. Second Round ?Write Formula 3. Semis ?Soluble Compound (compound that will dissolve in water) 4. Finals ?Soluble Product of Chemical Reaction 5.

### Sweet 16 Chemistry Compound Tournament

The result is the Sweet 16 Chemistry Compound Tournament. Hopefully, your students will enjoy playing the "Tournament" while reviewing these important topics. The rules for filling out the Tournament brackets are simple: For the first round, the student determines the chemical formula for the compound and counts the total number of atoms in the formula unit.

### Sweet 16 Chemistry Compound Tournament Key

Sweet 16 Chemistry Compound Tournament. Source(s): <https://shrink.im/a8z2K>. 0 0 0. Login to reply the answers Post; Caz. 5 years ago. This Site Might Help You. RE: How to do Sweet 16 chemistry Ion Tournament? The rule is to fill out the tournament brackets. For the first round you just add charge then.

### Sweet 16 Chemistry Ion Tournament

Sweet 16 Chemistry of Gases Tournament? First round: Predict the Gaseous product. Second round: Predict the one that dissolves in water to give an acidic solution. Semis: The gas that is more dense. Finals: The winner is a greenhouse gas produced in volcanic emissions. First round: Sodium Carbonate vs. Hydrochloric acid...

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### Sweet 16 Chemistry Ion Tournament

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First round ----- Predict the gaseous product. 1.) Na2CO3 vs. HCl 2.) Zn vs. HCl 3.) H2O2 vs. Yeast (catalyst) 4.) Na2SO3 vs. HCl 5.) Cu vs. HNO3 (concentrated) 6.) CaC2 vs. H2O 7.) NaOCl vs. HCl 8.) NH4NO2 vs. Heat 2nd round- The winner dissolves in H2O to give an acidic solution. 3rd round (semis) - The gas that is more dense (g/L) wins. Finals - The winner is a greenhouse gas produced in ...

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