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*KINETIC ENERGY • KINETIC ENERGY: Ability to do work as a result of the velocity of the system. • Energy associated with the velocity (v) of an object. • Example: A cool 1200 kg yellow car is running at 45 m/s. As a sharp turn is coming ahead the driver slows down to 20 m/s.*

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*STUDY GUIDE \_OATE CLASS use With Text Pages 214-221 Matter and Temperature Match the definition in Column II the term Column I. Write the letter Of correct definition in the blank on the left. Column I 1. kinetic theory Of matter 2. plasma 3. crystals 4. solid rzóamorphous material 6. steam \_e\_ 7. thermal expansion 8. liquid 10. ice*

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### Chapter 13 Study Guide Energy and Energy Resources

Energy, Work, and Power Test Study Guide – answer key. ... Weightlifter 2 does the same amount of work and uses less power than weightlifter 1. If kinetic energy of a 20kg mass is 160J, what is its speed? ( $KE = \frac{1}{2}m X V^2$ ) = ... Energy and Power Test Study Guide – answer key ...

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Study Guide and Reinforcement 3 ANSWER KEY 7. p! m " v! 2000 kg 10 m/s ! 20,000 kg • m/s 8. the 2000-kg truck because it has a greater mass Chapter 5, Energy Section 1 (p. 17) 1. d 7. second 2. d 8. decreases 3. d 9. unbalanced 4. a 10. remains the same 5. c 11. d 6. inertia Section 2 (p. 18) Note: Students' answers may be more or less

### Chapter 12 (States of Matter) Flashcards | Quizlet

84 Study Guide for An Introduction to Chemistry Section Goals and Introductions Section 7.1 Energy Goals To introduce the terms energy, kinetic energy, and potential energy. To introduce the Law of Conservation of Energy. To describe the relationships between stability, capacity to do work, and potential energy. To explain why breaking chemical bonds requires energy and why the formation of

### Physical Science Packet Chapter 16: Kinetic Theory of Matter

(1) the scent of cooking food spreads from the kitchen to the rest of the house; (2) the smell of a fart or turd quickly spreads from the bathroom to other parts of the house if the bathroom door is not closed or a candle is not lit right away; (3) a woman sprays perfume on when she goes to a nightclub, and the pleasant scent spreads to all the nearby males within seconds.

### Section/Objectives Standards Lab and Demo Planning

and distinguish between kinetic and potential energy. Heat flow is under stood in terms of ... Unit Length: Approximately 23 days Energy Unit Test Study Guide | Energy Unit Test Study Guide KEY Click on the links below for resources by Essential Question: EQ 1: How are potential and kinetic energy related ? ... Struggling students can do section 1

### 8th Grade Science Energy Unit Information

TEACHER GUIDE AND ANSWERS Chemistry: Matter and Change Teacher Guide and Answers 7 Study Guide - Chapter 12 – States of Matter Section 12.1 Gases 1. motion 2. a. small b. forces c. random d. elastic; kinetic 3.  $KE \frac{1}{2} mv^2$  4. Temperature 5. true 6. true 7. false 8. true 9. true 10. false 11. true 12. false 13. a 14. a 15. d 16. d 17. b 18. b ...

### Chapter 7 Energy and Chemical Reactions - Mark Bishop

standards. Differentiated Instruction Level 1 activities should be appropriate for students ... Analyze collisions to find the change in kinetic energy. Section 11.2 Section 11.1 Student Lab: Launch Lab, p. 285: basketball, metric ruler, ... Study Guide, pp. 9–14 Section 11-1 Quiz, p. 15 Teaching Transparency 11-1

### 13 STUDY GUIDE FOR CONTENT MASTERY 13 STUDY GUIDE FOR ...

Chapter 13 Study Guide Chapter Summary - Section 1: What is energy? 1. Energy is the ability to cause change. 2. A moving object has kinetic energy that depends on the object's mass and speed. 3. Potential Energy is energy due to position and depends on an object's mass and height. 4.

### Study Guide and Reinforce Answers - Hanover Area School ...

1 Physical Science Packet Chapter 16: Kinetic Theory of Matter ... Kinetic Theory of Matter Study Guide Major topics on the test will include: A. Kinetic Theory of Atoms a. What is an Atom? b. What is a Compound? c. Define the Kinetic Theory of Matter. ... Section 1 – Kinetic Theory & Section 2 – Properties of Matter . 17 Insert Solids ...

### Chapter 16 - Section 1 Review (Kinetic Theory) Flashcards ...

Study Guide for Content Mastery Chemistry: Matter and Change Chapter 13 73 States of Matter Section 13.1 Gases In your textbook, read about the kinetic-molecular theory. Complete each statement. 1. The kinetic molecular theory describes the behavior of gases in terms of particles in. 2. The kinetic-molecular theory makes the following ...

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