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properties of matter. Use each of the
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MASTERY Section 6.3 Periodic Trends
In your textbook, read about atomic radius and ionic radius. Circle the letter of the choice that best completes the statement or answers the question. 1. Atomic radii cannot be measured directly because the electron cloud surrounding the nucleus does not have a clearly defined c. outer edge.

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about calculating cell electrochemical
potential. Use the table of standard
reduction potentials below to answer
the following questions. 20. Suppose you
have two voltaic cells whose half-cells
are represented by the following

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Chapter 2 11 In your textbook, read about significant figures. Use each of the terms below just once to complete the statements. 7. The digits that are reported in an answer are called . 8. The numeral 9.66 has three significant figures, two known figures and one figure.

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your textbook, read about SI units.
Complete the following table. In your
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derived units. For each SI unit in
Column A, write the letter of the
matching item from Column B.

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SECTION 1.1 Earth Science In your
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of the major area of Earth science that
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meteorology geology ...

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reaction rate laws and determining

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reaction order. Use each of the terms below to complete the statements ...

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In your textbook, read about liquids and solids. In the space at the left, write true if the statement is true; if the statement is false,

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T182 Chemistry: Matter and Change

Name Date Class Name Date Class

CHAPTER 9 STUDY GUIDE FOR

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of the matching item in Column B.
Column A 1. The rays and particles that are emitted by a radioactive material

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