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The initial volume of a system containing 1 mole of an ideal gas

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at 292° Kelvin and 3 atm is 8 L. If the gas is cooled at constant volume until the pressure falls to 1.2 atm, then the gas is heated and expanded at constant pressure (1.2 atm) until the volume is 20 L and the temperature is 292° Kelvin, what is the overall work that has been done by the system?

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