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CHAPTER 8 SOLUTIONS MANUAL Covalent Bonding
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Chemistry: Matter and Change • Chapter 8 121
Section 8.1 The Covalent Bond pages 240–247
Practice Problems page 244 Draw the Lewis structure for each molecule. 1. PH_3 $\text{H}_2\text{C}=\text{CH}_2$ $\text{HC}\equiv\text{CH}$ respectively, for single, double, and triple P — — 2. H_2S $\text{H}_2\text{C}=\text{CH}_2$ $\text{HC}\equiv\text{CH}$ — H S S ...

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Section 9.1 Reactions and Equations pages 282–288
Practice Problems pages 284–287 Write skeleton equations for the following word equations. 1.

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• Chapter 6 87 CHAPTER 6 SOLUTIONS MANUAL

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Section 6.2 Classification of the Elements
pages 182–186 Practice Problems page 186 8.
Without using the periodic table, determine
the group, period, and block of an atom with
the following electron configurations. a.

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