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and Wood⁴ and of Smith-Magowan and
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the low temperature heat capacity data of
Tanner and Lamb⁶ which extend to high
concentration over the temperature range
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studied at 298.15 K by the method of
isothermal dissolution equilibrium. The
results indicated that the equilibrium solid
phase $\text{MgB}_4\text{O}_7 \cdot 9\text{H}_2\text{O}$ was kept in the
aqueous solution for 240 h and started to
transform into inderite
($\text{MgB}_3\text{O}_3(\text{OH})_5 \cdot 5\text{H}_2\text{O}$) after 240 h. The
concentration of the equilibrium liquid
phase with respect to ...

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