

## Sodium Sulfate Handbook Of Deposits Processing And Use

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Sodium sulfate is a typical electrostatically bonded ionic sulfate. The existence of free sulfate ions in solution is indicated by the easy formation of insoluble sulfates when these solutions are treated with Ba<sup>2+</sup> or Pb<sup>2+</sup> salts: Na<sub>2</sub>SO<sub>4</sub> + BaCl<sub>2</sub> → 2 NaCl + BaSO<sub>4</sub>. Sodium sulfate is unreactive toward most oxidizing or reducing agents.

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