

## Semiconductor 12th Cl Chapter Notes

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Trichloroethylene - Wikipedia

The strict definition of relative atomic mass is that  $A_r$  = average mass of all the isotopic atoms present in the element compared to 1/12th the mass of a carbon-12 atom. Example: chlorine consists of 75% chlorine-35 and 25% chlorine-37. So the relative atomic mass of chlorine is 35.5 or  $A_r(\text{Cl}) = 35.5$ . Ion Formation

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Platinum is a moderately hard metal (MH 3.5) of low mechanical strength, with a close-packed face-centred cubic structure (BCN 12). Compared to other metals in this category, it has an unusually high melting point (2042 K v 1338 for gold). Platinum is more ductile than gold, silver or copper, thus being the most ductile of pure metals, but it is less malleable than gold.

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Gist of the chapter Current Electricity 1. It is the rate of flow of charge 2. It is the work done per unit charge 3. It is additional small velocity imparted to an electron in the presence of electricfield. 4. It is drift velocity per unit electric field 5.

Metals close to the border between metals and nonmetals ...

Author(s) of the chapter, year of publication, and title of the chapter precede the details about the book editors and title of book. Following the word in is the editor(s) initials then family names. Enclose in brackets the edition number (if applicable). Enclose in brackets the page numbers of the chapter directly after the title. Schirm, V ...

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$\text{CH}_2 = \text{CH}_2 + \text{Cl}_2 \rightarrow \text{ClCH}_2\text{CH}_2\text{Cl}$ . When heated to around 400 °C with additional chlorine, 1,2-dichloroethane is converted to trichloroethylene  $\text{ClCH}=\text{CH}_2 + \text{Cl}_2 \rightarrow \text{ClCH}=\text{CCl}_2 + 3 \text{HCl}$ . This reaction can be catalyzed by a variety of substances. The most commonly used catalyst is a mixture of potassium chloride and aluminum chloride.

CBSE Class 12th Chemistry Notes: Solid State

p-type semiconductor: The semiconductor whose increased in conductivity is a result of electron hole is called a p-type semiconductor. When a crystal of group 14 elements such as Si or Ge is doped with a group 13 element such as B, Al, or Ga (which contains only three valence electrons), a p-type of semiconductor is generated.

NCERT Solutions for Class 12 Chemistry Chapter 1 The Solid ...

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