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Fundamentals 440. Another is reduction of an oxidized ion in solution by a so-called redox reaction. $O_2 + 4H^+ + 4e^- \rightarrow 2H_2O$. Assuming dissociation of water to H^+ and OH^- . Most corrosion reactions do involve water in either the liquid or condensed vapor phases, and is called the electrolyte.

PROF. DR. M.M. B. EL SABBAH AL-AZHAR UNIVERSITY FACULTY OF ...

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