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bipyramid formation is a
molecular geometry with one
atom at the center and 5
more atoms at the corners of
a triangular bipyramid. This

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is one geometry for which the bond angles surrounding the central atom are not identical (see also pentagonal bipyramid), because there is no geometrical arrangement with five terminal atoms in

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equivalent positions.

Trigonal bipyramidal
molecular geometry -
Wikipedia

In a tetrahedral molecular
geometry, a central atom is
located at the center with

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four substituents that are located at the corners of a tetrahedron. The bond angles are $\cos^{-1} \left(\frac{1}{3} \right) = 109.4712206...^\circ \approx 109.5^\circ$ when all four substituents are the same, as in methane (CH_4) as well as its

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heavier analogues. Methane
and other perfectly
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Tetrahedral molecular
geometry - Wikipedia

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(Z) , the number of protons in the nucleus, and a mass number (A), the total number of protons and neutrons in the nucleus.

Nuclide, Atomic Number, mass number - Chemistry

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