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Summary 2.1

Matter An element is a substance that cannot be broken down into simpler substances by chemical or physical means. An atom is the smallest particle of matter that contains the characteristics of an element. □ The

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central region of an  
atom is called the  
nucleus.

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people make  
scientific to  
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information,  
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used to make  
coins? 9. What is  
the benefit of  
pumping nitrogen  
gas into seawater  
that is stored in

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steel tanks?  
Recognizing  
Chemical Changes  
(pages 56–57) 10.

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the need. Research

the problem.

Design a solution.

Build a prototype.

Troubleshoot and

redesign.

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2 11 Matter

Substance Element

Compound

Homogeneous  
mixture

Heterogeneous  
mixture Mixture

true Elements true

An element has a  
fixed composition  
because it contains



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only one type of  
atom. Compounds  
An element is a  
substance that  
cannot be broken  
down into simpler  
substances. b c a

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8 67 Name \_\_\_\_\_

Chapter 8 Class

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Solutions, Acids,  
and Bases Factors  
Affecting Solubility  
(page 237) 7. Circle  
the letters of  
factors that affect  
the solubility of a  
solute. a.

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Chapter 12 Forces  
and Motion Friction  
(pages 359-360)

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following sentence  
true or false? ...

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Date 7. Your in-line  
skates are sitting in  
a box on ...

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Energy Resources.  
Chapter 15 Energy  
136. The DRL  
should be  
highlighted in

Chapter  
2 Properties of  
Matter Section 2.3  
Chemical  
Properties  
Is the following  
sentence true or  
false? One

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potential danger of radon gas is that prolonged exposure to it can lead to lung cancer.

16. Detecting Nuclear Radiation (page 297)

Name two devices that are used to detect nuclear radiation.

- Geiger counters
- Film



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10.1 Radioactivity

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Answers

a. Petroleum accounts for most of the energy produced in the world. b. Petroleum fuels most cars, airplanes, trains, and ships. c. The United States consumes a third of the oil produced in the world. d. Three percent of the world's oil

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