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In the equation, ${}_{6}^{14}\text{C} \rightarrow {}_{7}^{14}\text{N} + {}_{-1}^{0}\text{e}$, the _____ decay of radioactive carbon-14 results in the creation of a new nitrogen-14 atom. answer choices
Radioactive

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key chemistry as ... Write the nuclear
equation for this reaction and identify the
two other particles. $n_0^1 + {}_{92}^{235}\text{U} \rightarrow {}_{52}^{137}\text{Te} + {}_{40}^{97}\text{Zr} + 2n_0^1$
The two other
particles formed are neutrons.

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Key Takeaways Key Points. A balanced nuclear equation is one where the sum of the mass numbers (the top number in notation) and the sum of the atomic numbers balance on either side of an equation. Nuclear equation problems will

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often be given such that one particle is missing.

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equations is important when trying to understand nuclear reactions. All equations need to be balanced to conform to two conservation laws: the mass number is conserved, and the electrical charge is conserved. Success Criteria Use the conservation laws to find an unknown in a nuclear reaction equation. Write a balanced nuclear equation for a ...

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Choose the correct answer for each
question Write the letter of the answer on
the line to the right of the equation You
will need a periodic chart to do this
worksheet Identify the missing particle in
the following nuclear

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Understanding nuclear reactions is
important not only for engineers, but also

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doctors, nurses, and environmental professionals. In this module you will:
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