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$1/10^9 \text{ m} = 1 \text{ nm}$ $1 \text{ m} = 10^9 \text{ nm}$ Relative Sizes
Radii (in m) Example 10^{-10} Atom of hydrogen
 10^{-9} Molecule of water 10^{-8} Molecule of
haemoglobin 10^{-4} Grain of sand 10^{-2} Ant 10^{-1}

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Watermelon Fig. 3.2: An image of the surface of silicon 3.2.1 WHAT ARE THE MODERN DAY SYMBOLS OF ATOMS OF DIFFERENT ELEMENTS? Dalton was the first scientist to ...

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the vacant anionic sites, creating F-centres.

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