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Pogil Mole Ratios Worksheet Answers
MODEL 1: Mole-to-Mole Stoichiometry (1-step) Example: mole-to-mole Question: Given the following equation: wod be 1N2 (g)+3 H2 (g) 2 NH₃ (g) How many moles of H₂ are needed to produce 34.8 moles of NH₃? Mole ratio Mole ratio Answer: 34.8 mol NH₃ x 3 mol H₂/2mol NH₃ = 52.2 mol H₂ OR Key Questions: 1.

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