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Module Six - Limiting Reagents, Theoretical

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Yields and Percent Yields where the actual and theoretical yields may be expressed in grams or moles (provided that the same unit is used for each). **Example 5.** Nitrobenzene, $C_6H_5NO_2$, is an important raw material for the preparation of dyes. It is prepared from benzene, C_6H_6 , and nitric acid, HNO_3

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Biochemistry Page 2 Module Six - Limiting
Reagents, Theoretical Yields and Percent Yields
Determining the Limiting Reagent and Excess
Reagent One of the methods used to synthesize
urea, $(\text{NH}_2)_2\text{CO}$, is to react ammonia, NH_3 ...**

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Module Six - DePauw

**See, also, Problems 1, and 3-10 in Module 5.
DePauw University - Department of Chemistry
and Biochemistry Page 6 Module Six - Limiting**

Reagents, Theoretical Yields and Percent Yields where the actual and theoretical yields may be expressed in grams or moles (provided that the same unit is used for each). Example 5.

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calculate the initial response of a rectangle located at A and a rectangle located at B. Then, calculate the total amount of inhibition imposed on both rectangles from their surround. Finally, calculate the final output of both rectangles after lateral inhibition. The following rules apply: Areas illuminated by WHITE gives off an intensity of 100 Areas illuminated by BLACK gives of an ...

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Atoms, Molecules and Moles - DePauw - DePauw University

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Then calculate the total amount of inhibition imposed on ...

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Slide 1

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and Biochemistry Page 5 Module Two - Atoms, Molecules and Moles mass units (amu), where 1 amu is defined as 1/12 of the mass of an atom of carbon-12.

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Stoichiometric Calculations Module Six Limiting Reagents, Theoretical Yields, and Percent Yields DePauw University - Department of Chemistry and Biochemistry Page 1

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