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Attraction between metal atoms and the surrounding sea of elec... Chemical Bond Forces that hold groups of atoms together and make them functi... Ionic Bonding Chemical bonding that results from the electrical attraction b... A chemical formula that shows the number and kinds of atoms in... The energy required to break the bonds in one mole of a chemic....

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b. the same as covalent bonding. c. a combination of ionic and covalent bonding. different from ionic or covalent bonding. 2. he valence electrons in a metallic bond move freely throughout the network of metal atoms. b. are held tightly by the most positively charged atom. c. are shared equally between two metal atoms.

6 Chemical Bonding - Effingham County Schools / Overview

Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued ____13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. ____14. When there is a small decrease in temperature, the average kinetic energy of the particles of a liquid

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a. Meyer c. Stas b. Ramsay d. Moseley ____ 13. The most useful source of general information about the elements for anyone associated with chemistry is a. a. calculator. c. periodic table. b. table of metric equivalents. d. table of isotopes. ____ 14. The periodic table. a. permits the properties of an element to be predicted before the element ...

Assessment Chapter Test B

Modern Chemistry 88 Chapter Test Chapter: States of Matter PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. ____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide. b. repel each other and collide.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

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b. Br⁻, bromide ion. c. Br⁺, bromium ion. d. Br⁻, bromium ion. 2. The platinum(II) ion and the platinum(IV) ion a. are anions. b. are polyatomic ions. c. have charges of 2+ and 4+, respectively. d. have charges of 1+ and 3+, respectively. 3. is the name of the compound made of zinc ions, Zn and

fluoride ions, F — ? a. zinc difluoride b. zinc fluorate

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Explain your answer. The three oxygen atoms have oxidation states of 6 total, and because the algebraic. sum of the oxidation states in a neutral compound must be zero, the two nitrogen. atoms must have oxidation states of 6 total, therefore 3 each.

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CH 1 Reading Assignment Modern Chemistry. CH 1 Vocabulary-New. CH 1 Mixed Questions. CH 1 Matter & Energy Vocabulary-New. CH 1 Classifying Matter-New. CH 1 Classification of Matter ... CH 11 Pre Test Answers. CH 11 Section Review 11-1 - 11-3 . CH 11 Gas Stoichiometry 1 . CH 11 Gas Stoichiometry 2. CH 11 Boyles Law . CH 11 Charles Law ...

Modern Chemistry Test B Answer

Assessment Chapter Test B - clarkchargers.org. Modern Chemistry 71 Chapter Test Name Class Date Chapter Test B, continued 17. A substance combines with oxygen, releasing a large amount of energy as heat and light, in a(n) . 18. The decomposition of a substance by an electric current is called . 19.

7 Chemical Formulas and Chemical Compounds

Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. b In metals, the valence electrons are considered to be (a) attached to particular positive ions. (c) immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2. a The fact that metals are malleable and ionic crystals are brittle is best

Assessment Chapter Test A

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Modern Chemistry Chapter Test B Answer Key

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Name CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9, If a particular ompound is composed of elements A and B, the ratio of the mass of B to the mass of A will alw ys be the same.

7 Chemical Formulas and Chemical Compounds

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Assessment Chapter Test B - Ed W. Clark High School

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an

atom in a chemical compound to attract electrons from another atom in the compound is called _____. 17. The energy required to remove one electron from an atom is called its _____. 18.

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