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you require to
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needs later than
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significantly
cash? Why don't
you try to
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something basic
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Table and*

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Periodic Law

Section 6.1

*Development of
the Modern*

Periodic Table

*1. octaves 2.
eight 3. nine 4.
accepted 5.*

Dmitri Mendeleev

6. atomic mass

7. elements 8.

atomic number 9.

Henry Moseley

10. protons 11.

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proposed to
explaination of
the
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radiation in
whole numbers of
photons and in
order for an
electron to be
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CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer

the following
questions in the
space provided.

1. c In the
modern periodic
table, elements

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Answer
are ordered (a)
according to
decreasing
atomic mass. (b)
according to
Mendeleev's
original design.
(c) according to
increasing
atomic number.
(d) based on
when they were
discovered. 2. d
Mendeleev

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certain*

*similarities in
the ...*

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15 When Electron

egativities of

bonding atoms

are the same (as

they are in

diatomic

molecules) or

close to the

same, they SHARE

electrons. Bonds

formed when

atoms share

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*electrons are
called Covalent
Bonds .*

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CHAPTER 10

*REVIEW States of
Matter SECTION 3*

SHORT ANSWER

*Answer the
following
questions in the*

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space provided.

Answer
1. Match

description on
the right to the
correct crystal
type on the
left. b ionic
crystal (a) has
mobile electrons
in the crystal c
covalent
molecular
crystal (b) is
hard, brittle,

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and

Answer

*nonconducting a
metallic crystal
(c) typically
has the lowest
melting point of
the four*

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a mutual

electrical

attraction

between the

nuclei and the

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valence

Answer

*electrons of
different atoms
that binds the
atoms together*

*10 States of
Matter - Ms.*

Agostine's

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*The rest of the
pages are the
exact same as
the non state*

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