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Modern Chemistry 1 Arrangement of Electrons in Atoms CHAPTER 4 REVIEW Arrangement of Electrons in Atoms Teacher Notes and Answers Chapter 4 SECTION 1 SHORT ANSWER 1. In order for an electron to be ejected from a metal surface, the electron must be struck by a single photon with at least the minimum energy needed to knock the electron loose. 2.

CHEMISTRY CHAPTER 4 REVIEW - Holt Modern Chemistry Review ...

Taken from the book Modern Chemistry by Holt, Rinehart, and Winston on Chapters 4 and 5, which deals with electrons and the periodic table. Includes the chapter vocabulary and a few other useful things.

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CHAPTER 4 REVIEW Arrangement of Electrons in Atoms

This scientist proposed to explanation of the photoelectric effect is electromagnetic radiation in whole numbers of photons and in order

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. c

In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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SECTION 1 continued 6. Some binary compounds are ionic, others are covalent. The type of bond favored partially depends on the position of the elements in the periodic table.

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4. What two factors determine whether or not a molecule is polar?  
electronegativity difference and molecular geometry or unshared  
electron pairs 5. Arrange the following types of attractions in order of  
increasing strength, with 1 being the weakest and 4 the strongest. 3  
hydrogen bonding 4 ionic 2 dipole-dipole 1 London dispersion 6.

7 Chemical Formulas and Chemical Compounds  
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6 Chemical Bonding

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