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physical changes.

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT

ANSWER Answer the following questions in the space provided.

a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

2 Measurements and Calculations

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT

ANSWER Answer the following questions in the space provided.

Match description on the right to the correct crystal type on the

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b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metal crystal (c) typically has the lowest melting point of the four

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter
SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode. The smallest unit of an element that can exist either alone or in molecules containing the

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6 Chemical Bonding

3. Compare and contrast a model with a theory. Theories are broad generalizations used to explain observations. Models are a physical object used to illustrate or explain complex concepts or an explanation of how phenomena occur and how data and events are related.

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3 Atoms: The Building Blocks of Matter

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter

SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. In cathode-ray tubes, the cathode ray is emitted from the negative electrode, which is called the cathode. The smallest unit of an element that can exist either alone or in molecules containing the

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3:3 mol H₂; or their reciprocals 7. Given the following equation
 $4\text{NH}_3(\text{g}) + 6\text{NO}(\text{g}) \rightarrow 5\text{N}_2(\text{g}) + 6\text{H}_2\text{O}(\text{g})$ 1 mol NO:1 mol H₂O a.
What is the mole ratio of NO to H₂O? 3 mol NO:2 mol NH₃ b.
What is the mole ratio of NO to NH₃? 0.360 mol c. If 0.240 mol
NH₃ react according to the above equation, how many moles of
NO will be consumed? 8.

Modern Chemistry Chapter 2-3 Review Answers

SECTION 1 continued 6. Some binary compounds are ionic, others
are covalent. The type of bond favored partially depends on the
position of the elements in the periodic table.

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MODERN CHEMISTRY The atomic number of nickel-60 is 28. How

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many neutrons does this iso- tope have? Carbon-14 has 8 neutrons. What is the atomic number of carbon-14? 11 CHAPTER 3 TEST ANSWERS HR W material copyrighted under notice appearing earlier in this work.

7 Chemical Formulas and Chemical Compounds

CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT

ANSWER Answer the following questions in the space provided.

In the modern periodic table, elements are ordered (a) according to increasing atomic mass. (b) according to Mendeleev's original design. ... 40 THE PERIODIC LAW MODERN CHEMISTRY ...

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5 The Periodic Law

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