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12H 22O 11 10 atoms c.  $\text{Fe}(\text{ClO}_2)_3$  9 atoms d.  $\text{Fe}(\text{ClO}_3)_2$  5. N 2O 5 a. What is the formula for the compound dinitrogen pentoxide? iron(II) oxide b. What is the Stock system name for the compound  $\text{FeO}$ ? H 2SO 3 c. What is the formula for

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sulfurous acid? phosphoric acid d. What is the name for the acid  $H_3PO_4$ ? MODERN CHEMISTRY CHEMICAL FORMULAS

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HOMEWORK 12-3 (pp. 273–375) ...  
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work. Name Date Class . SECTION 12-3  
continued b. Explain the difference  
between an open system and a closed  
system. 5.

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CHAPTER 5 REVIEW The Periodic Law  
SECTION 1 SHORT ANSWER Answer  
the following questions in the space  
provided. 1. c In the modern periodic  
table, elements are ordered (a) according  
to decreasing atomic mass. (b) according  
to Mendeleev's original design. (c)  
according to increasing atomic number.  
(d) based on when they were discovered.

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