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MIXED PROBLEMS Name What volume of NH_3 at STP is produced if 25.0 g of N_2 is reacted with an excess of $0.8008 \text{ L } \text{OKC103} \cdot \text{POKCl} + \text{gO}_2$... MOLE-MOLE PROBLEMS I. $\text{N}_2 + 3\text{H}_2$ Name ... MASS-MASS PROBLEMS 1. $2\text{KC103} + \text{ }_302$ Name

Mixed Stoichiometry Worksheet Answers

Mixed Mole Problems Directions: Solve the following problems. 1) How many grams are there in 1.5×10^{25} molecules of CO_2 ? 2) What volume would the CO_2 in Problem 1 occupy at STP? 3) A sample of NH_3 gas occupies 75.0 liters at STP. How many molecules is this? 4) What is the mass of the sample of NH_3 in Problem 3?

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Moles Stoichiometry Mole Mole Problems Answers if8766 page 50. Mixed Mole Problems of atoms of oxygen are in the container Chemistry IF8766 Page 53. [FREE] Mixed Mole Problems Worksheet Answers Two sulfuric acid solutions are mixed as follows: 25.0 mL of a 0.50 M sulfuric acid solution Page 1/3

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