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The kinetic molecular theory is a simple but very effective model that effectively explains ideal gas behavior. The theory assumes that gases consist of widely separated molecules of negligible volume that are in constant motion, colliding elastically with one another and the walls of their container with average velocities determined by their absolute temperatures.

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temperature and molecular motion? 6. In terms
of the kinetic-molecular theory of gases, how
can increase in the temperature of a gas
confined in a rigid container cause an

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Kinetic Molecular Theory Kinetic Molecular
Theory (KMT) allows us to explain the
observed macroscopic (measurable) properties
of ideal gases (pressure, volume,
temperature, etc.) in terms of the
microscopic properties of individual gas
molecules (or atoms, for monatomic gases). As
we will see later, real gases do not follow

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KMT under all conditions, leading to deviations from the ideal gas law.

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Authored by Applications 1. There is a government warning on all aerosol cans that states: Do not store at a temperature above 120° F (50°C). a) Explain why this warning is required in terms of the relationship between temperature and pressure and the kinetic molecular theory.

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POGIL: Kinetic Molecular Theory Learning Objectives Identify the basic differences between particle behavior in the solid, liquid, and gaseous phases. Develop an understanding of the postulates of the kinetic molecular theory. Identify how temperature affects molecular motion.

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In terms of the kinetic molecular theory of gases, explain how increasing the temperature of a gas confined in a rigid container causes an increase in the pressure of the gas?

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Applications 14. There is a government warning on all aerosol cans that states: Do not store at a temperature above 120°F (50°C).

Ms. Demonte's Chemistry Classes - Home Name_____ Kinetic Molecular Theory Why? The kinetic-molecular theory is a model or a mental image of how particles of matter behave. Knowledge of the kinetic-molecular theory allows us to predict the action of solids, liquids and gases and understand how the changes of state occur. Learning Objectives: Identify the basic differences

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Kinetic Molecular Theory Applications 1.

There is a government warning on all aerosol cans that states: Do not store at a temperature above 1200 F (5000. a) Explain why this warning is required in terms of the relationship between temperature and pressure and the kinetic molecular theory. hljhhq -flu (enja//l.u- more QUjctI prrjw<-

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