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1) Use your knowledge of ionic equations to give the molar ratio in which the following acids react with bases. Complete the table to show your answers. Acid Formula of acid Base Formula of base

### 3.1.6 Chemical equilibria, Le Chatelier's principle and Kc ...

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-1 k-1. 8 25 cm<sup>3</sup> of 2.0 mol dm<sup>-3</sup> nitric acid was reacted with 25 cm<sup>3</sup> of 2.0 mol dm<sup>-3</sup> potassium hydroxide in an insulated cup. The temperature rose from 20.2°C to 33.9°C.

### Calculating Equilibrium Constant Kc | Teaching Resources

Kc Calculation? In combustion studies of H<sub>2</sub> as an alternative fuel, you find evidence that the hydroxyl radical (HO) is formed in flames by the following reaction.  $\text{H(g)} + \frac{1}{2} \text{O}_2\text{(g)} \rightleftharpoons \text{HO(g)}$

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