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curriculum at MIT was
offered in 1888 and
helped to establish
chemical engineering
as a discipline. Since
then, members of the
MIT Department of
Chemical Engineering
have developed the*

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tools and guidelines
to define and advance
the field.

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If a process has to be
done in batches,
several reactors are
often used in parallel,
shifted in time to give
a continuous stream
from the group of

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*reactors. See Figure 3
for a schematic
representation of a
CSTR. 2.3 Plug ow
reactor(PFR) Another
type of continuous
reactors is the plug
ow reactor, or PFR.*

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*Chemical Processes:
Principles, Analysis,
Synthesis enhances
student understanding
of the connection
between the
chemistry and the
process.*

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engineering analysis
of physical and
chemical processes.*

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concepts of rate
processes, energy
and mass transport,
and kinetics of
chemical reactions.*

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offered by the
Stanford University
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Department. It
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overview of the
chemical engineering
field today...

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“Chemicalengineers”
use math, physical
sciences (physics,
chemistry), life
sciences (biology,
microbiology and
biochemistry), and*

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economics to solve practical problems.
The difference between chemical engineers and other types of engineers is that they apply a knowledge of chemistry in addition to other engineering disciplines.

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Prerequisites o 1.1

Consistency of units

1.1.1 Units of

Common Phys

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quantitative treatment
of differential
equations that arise in
the area of chemical
engineering. The
book emphasizes the
methods implemented
in commercial
software to help the
engineer gain*

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maximum benefit.

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*Engineering - Home
Chemical engineering
is a branch of
engineering that uses
principles of
chemistry, physics,
mathematics, biology,
and economics to
efficiently use,
produce, design,*

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transport and transform energy and materials. The work of chemical engineers can range from the utilisation of nano-technology and nano-materials in the laboratory to large-scale industrial processes that convert chemicals, raw materials, living cells, microorganisms,

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and energy into useful forms and products.

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Problem

*considerations with
molecular balances.*

Degrees of Freedom.

*Independent and
dependent chemical
reactions. Inerts*

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*versus Reactive
Species. Equilibrium
constants
(introduction/review
from general chem)
Extent of Reaction is
still Extent of
Reaction. Example
Problem without
equilibrium. Example
Problem with
equilibrium.*

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*DESCRIPTION The
goal of this book is to
help the student
experience chemical
engineering to the
fullest extent possible*

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*within the constraints
of limited time and*

limited student

*background. In pursuit
of that goal, it teaches
the freshman to solve
quantitative problems,
although at a low level
of complexity and
within a scope that is
narrow and well-
defined.*

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Prerequisites o 1.1
Consistency of units
1.1.1 Units of*

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*Common Physical
Properties 1.1.2 SI
(kg-m-s) System*

*1.1.2.1 Derived units
from the SI system*

*1.1.3 CGS (cm-g-s)
system*

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To Chemical processing industry is a prerequisite course in chemical engineering. is the conversion or changing of a chemical compound. This conversion can be spontaneous or by an external agent. It involves chemical reaction.

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chemical engineers
concern themselves
with the chemical
processes that turn
raw materials into
valuable products.*

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- Wikipedia

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