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## **1.1 Chemistry 1**

**Chapter 13 Interpreting Graphics and Vocabulary Name . Part A - Intermolecular Forces. 1. Fill in the diagram (with high or low) to show how intermolecular forces influence the volatility, vapor pressure, and boiling point of a substance.**

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[hs.pequannock.org](http://hs.pequannock.org)**

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Answers**

**The following is on Chapter 14 Interpreting  
Graphics. Trial 1 Mass of Flask + stopper =  
82.32 grams Mass of flask + stopper  
+ condensed vapor = 83.73 g Temperature of  
boiling water = 99 degrees celsius Barometric  
Pressure 773.5 mm Hg Trial 2 mass of  
flask+stopper= 83.39 grams Mass of flask +  
stopper + condensed vapor = 83.82 grams  
Temperature of boiling water = 99 degrees  
celsius Barometric ...**

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Graphics ...**

**Chapter 11 Small-Scale Lab Section 11.3  
Precipitation Reactions: Formation of Solids,  
page 345 Analysis 1. a.  $\text{Na}_2\text{CO}_3 + 2\text{AgNO}_3 \rightarrow$   
 $2\text{NaNO}_3 + 3\text{Ag}_2\text{CO}_3(\text{s})$  b.  $2\text{Na}_3\text{PO}_4 + 3\text{Pb}(\text{NO}_3)_2 \rightarrow$   
 $6\text{NaNO}_3 + \text{Pb}_3(\text{PO}_4)_2(\text{s})$**

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Review, Interpreting Graphics ...**

**chem\_TE\_ch25.fm Page 803 Thursday, April 20,  
2006 8:47 AM. 804 Chapter 25 Section 25.2  
(continued) Half-Life Discuss Explain that, for**

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***each element, there exists only a small range of neutron-to-proton ratios that produce stable nuclei. If a nucleus does not reflect a***

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10.3***

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***Assuming salicylic acid  $C_7H_6O_3$  as the limiting  
reagent calculate the theoretical yield. Molar  
mass of  $C_7H_6O_3 = 132.1$  g/mol. Molar mass of  
aspirin = 180.1 g/mol***

**Prentice Hall Chemistry Chapter 13 Interpreting  
Graphics ...**

**Chapter 11 Chemical Reactions Interpreting  
Graphics 1 . Name Date Class carbonic acid \_\_\_\_\_**

**nitro benzene nitric acid ... 2 Chapter 11  
Chemical Reactions Interpreting Graphics.**

**Name Date Class CHEMICAL REACTIONS**

**Vocabulary Review Each clue describes a  
vocabulary term. Read the clues and write the  
letters of each term on the lines provided.**

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Introduction to Chemistry 7 1.1 FOCUS**

**Objectives 1.1.1 Identify five traditional areas  
of ... Review, Interpreting Graphics ... 8 Chapter  
1 Areas of Study Because the scope of  
chemistry is vast, chemists tend to focus on one  
area.**

**10.3 Percent Composition and Chemical  
Formulas 10**

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Spell. Test. PLAY. Match. Gravity. Created by.  
madigil33. Terms in this set (35) Figure 33 If P1  
is 50 kPa and P2 is 100 kPa, what are V1 and  
V2? 2.0L 1.0L. How would the graph in Figure  
3-3 change if the volume of the cylinder  
remained constant and temperature was the  
manipulated variable?**

**25.2 Nuclear Transformations 25**

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