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Chapter 21: Hydrocarbons

2 A Self-Study Guide to the Principles of Organic Chemistry 6.5

Electron Configuration of the Elements Commonly Involved in

Organic Compounds Figure 1.1 Hydrocarbon, methane and its

derivatives The Periodic Table contains more than one hundred

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elements; however, organic chemistry deals mostly with only certain elements.

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13.1 CLASSIFICATION

Hydrocarbons are of different types. Depending upon the

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Chemistry Aswers

Study Guide Key Concepts 22.1 Hydrocarbons □ Because carbon has four valence electrons, carbon atoms always form four covalent bonds. □ The carbon atoms in an alkane can be arranged in a straight chain or in a chain that has branches. □ Molecules of hydrocarbons, such as alkanes, are nonpolar molecules. 22.2 Unsaturated Hydrocarbons

Quiz: Hydrocarbons

Total combustion of pure hydrocarbons produces only carbon dioxide and water. However, real combustion engines use fuel with numerous impurities and don't always break down hydrocarbons completely...

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from only carbon and hydrogen atoms. Such hydrocarbons are the
simplest organic compounds, but they are also of prime economic
importance because they include the constituents of petroleum and

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natural gas.

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