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synthesis of
hydrazine
derivatives from
aldehydes ...
Phenylhydrazine
was the first

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hydrazine
derivative
characterized,
reported by
Hermann Emil
Fischer in 1875.

[6] [7] He
prepared it by
reduction of a
phenyl diazonium
salt using
sulfite salts.
Fischer used
phenylhydrazine

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(G.M et al 2010)
with slight modi
fication. 32.5
gm. of ethylacet
oacetate was
taken in 250 ml
conical flask
and stirred
magnetically

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a solution of
hydrazine
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gm. hydrazine
hydrate in 20 ml
ethanol). The
temperature of
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mixture was
regulated at
60°C ...

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*been
considerable
development in
this field and
many new aspects
of hydrazine
chemistry and
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have evolved.*

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The hydrazine scaffold is a common skeleton in biologically active compounds. 1

Hydrazine derivatives are frequently employed as precursors of various heterocyclic compounds. 2

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They also serve as protected amines, which are transformed to amines by N-N bond cleavage. 3

However, hydrazines, including structurally simple alkyl hydrazines are difficult to synthesize

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directly from
hydrazine ...

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Electrochemical
oxidation of
hydrazine and
its derivatives
on the surface
of various metal
electrodes in

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*alkaline media
was*

investigated. A

comparison of

various

polycrystalline

metal electrodes

(Ni, Co, Fe, Cu,

Ag, Au, and Pt)

showed that Co

and Ni

electrodes have

a lower onset

potential for

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Properties.

*hydrazine
oxidation than
the Pt
electrode.*

Applications

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: Preparation,
Properties ...
Hydrazine is a
precursor to
several
pharmaceuticals
and pesticides.*

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Often these applications involve conversion of hydrazine to heterocyclic rings such as pyrazoles and pyridazines. Examples of commercialized bioactive hydrazine derivatives

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include
cefazolin,
rizatriptan,
anastrozole,
fluconazole,
metazachlor,
metamitron,
metribuzin,
paclobutrazol,
diclobutrazole,
propiconazole,
hydrazine ...

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Pyrazole and its

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derivatives are prepared by dehydrogenating 2-pyrazoline or its derivatives by a process in which the reaction is carried out using sulfuric acid in the presence of iodine or of an iodine compound

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at from 50° to
 250° C.

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Synthesis of
Some Pyrazolone
Derivatives and
Evaluation of

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The Knorr
pyrazole
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organic reaction
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a hydrazine or

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Phenylhydrazine

– Wikipedia

Hydrazine

(diamide), N_2H_4 , a colorless liquid having an ammoniacal odor, is the simplest diamine and unique in its class because of

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*the N N bond.
Hydrazine and
its simple
methyl and
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derivatives have
endothermic
heats of
formation and
high heats of
combustion.
Hence, these
compounds are
used as rocket*

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of hydrazine
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