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process: (1) changes in a system, (2) cycles in a system, and (3) flow processes. (1) A change in a system is defined by a passage from an initial to a final state of thermodynamic equilibrium. In classical thermodynamics, the actual course of the process is not the primary concern, and often is ignored.

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chemical, and physical
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Chemical shift of peak f
- chemical shift of peak
 $g = 1.2470 \text{ ppm} -$
 $1.1293 \text{ ppm} = 0.1177$

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Chemical Equilibrium

ppm Chemical shift of
peak e - chemical shift
of peak f = 1.3646 ppm

- 1.2470 ppm = 0.1176

ppm Average = 0.11765

ppm $0.11765 \text{ ppm} \times 60$

$\text{MHz} = 7.059 \text{ Hz}$? $J(\text{H}-$

$\text{H}) = 7.06 \text{ Hz}$ Notice the

coupling constant for

this multiplet is the

same as that in the

example.

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