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(ii) Among CH_3NH_2 and $(\text{C}_2\text{H}_5)_2\text{NH}$, primarily due to the greater +I-effect of the two $-\text{C}_2\text{H}_5$ groups over one $-\text{CH}_3$ group, $(\text{C}_2\text{H}_5)_2\text{NH}$ is more basic than CH_3NH_2 . In both $\text{C}_6\text{H}_5\text{NH}_2$ and $\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2$ lone pair of electrons present on N-atom is delocalized over the benzene ring but $\text{C}_6\text{H}_5\text{N}(\text{CH}_3)_2$ is more basic due to +I effect of two $-\text{CH}_3$ groups.

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