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Cr +2 TOTAL -14 9. 10. 11. [a]  
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oxidation state is always +1 [b] oxygen  
usually has an oxidation state of—2 [C]  
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is the product of the subscript and the  
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Ca(OH)<sub>2</sub> + 3242 25. What is the mass of  
4.2 moles of H<sub>3</sub>P<sub>0</sub>4? 26. What is the mass  
Of 0.75 mole of butane, C<sub>4</sub>H<sub>8</sub> 911,6 47  
50 27. What is the mass of 2 moles of  
gypsum, CaC<sub>12</sub>.2H<sub>2</sub>O ? 2HzO 2(35) 4

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