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Describe how iodine/thiosulfate titrations are carried out: -weigh solid oxidising agent and dissolve in water to make 250cm³ (or use aqueous hydrogen peroxide as the oxidizing agent). 25cm³ portions of the oxidizing agent solution are pipetted into a conical flask and some sulfuric acid is added.

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Unit 3 IAL Edexcel Chemistry. Sulfur dioxide evolved and pale yellow precipitate formed after adding dilute sulphuric or hydrochloric acid.

