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Problems for

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interactive quiz. Use the worksheet to identify study points to watch...

Dilutions Worksheet
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Dilution Problems
Worksheet ($M_1 V_1 = M_2 V_2$) 1. How much of a 15.0 M stock solution do you need to prepare

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250 ml of a 2.35 M HF solution? 2. If 455 ml of 6.0 M HNO_3 is diluted to 2.5 L, what is the molarity of the diluted solution?

Dilutions Worksheet

Name: KEY -

Chemistry 301

Microbiology BIOL

275 Dr. Eby Bassiri e

bassiri@sas.upenn.

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edu 3 dilution.

Finally add 0.1 ml of the 10^{-6} dilution to 0.9 ml of diluent to make 1 ml of 10^{-7} dilution. We have used just 1 microliter of the sample and 2898 microliters of diluent.

Dilution Problems

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Answer Key

Dilution Problems

#1 - 10. Return to Solutions Menu. Ten Examples. Problems #11 - 25. Problems #26 - 35. Problem #1: If you dilute 175 mL of a 1.6 M solution of LiCl to 1.0 L, determine the new concentration of the solution.

Solution: ... That

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way, x is the answer
you want, ...

Molarity and Serial
Dilutions Teacher
Handout

Placing the proper
values into the
dilution equation
gives: (2.500 mol/L)
 $(100.0 \text{ mL}) = (0.5500$
 $\text{mol/L}) (x) \quad x = 454.5$
 mL Sometimes the
problem might ask

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Key

how much more water must be added. In this last case, the answer is $454.5 - 100.0 = 354.5$ mL. Go ahead and answer the question, if your teacher asks it, but it is bad technique in the

DILUTION

EQUATION - Los

Page 14/34

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Key
Alamitos Unified
School District

Practice Problems:
Solutions (Answer
Key) 1. What mass
of solute is needed
to prepare each of
the following
solutions? a. 1.00 ...

Practice Problems:
Solutions (Answer
Key)

This is a chemistry

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tutorial that covers dilution problems, including examples of how to calculate the new concentration of a diluted solution, and how to ca...

Molarity Problems

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EQUATION Unit 16.4

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Chapter 15 Section 4

Pg. 569-577.

Changing

concentrations

DILUTION

PROBLEMS

ANSWER KEY PDF -

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-Key vocabulary will

be introduced to the

class as a whole-

Everyone will work

example

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dimensional

analysis problems

together-Students

will break into small

groups to do the

serial dilution

activity with Kool-

Aid-Students will

discuss and

respond to

questions on the

handout-Students

will do the

mole/molarity Nerd

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Problems Answer

activity in small
groups

4: Dilution

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dilution problems
answers.

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Key
Dilution Problems,
Chemistry, Molarity

& Concentration

Examples, Formula

& Equations

Dilutions Worksheet

W 329 Everett

Community College

Student Support

Services Program 1)

If 45 mL of water are

added to 250 mL of

a 0.75 M K_2SO_4

solution, what will

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Problems Answer

Key
the molarity of the diluted solution be?

2) If water is added to 175 mL of a 0.45 M KOH solution until the volume is 250 mL, what

Dilutions Worksheet

- nclark.net

To work the problem, you need 3 values---a colony count from the pour

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or spread plates, a dilution factor for the dilution tube from which the countable agar plate comes, and the amount of the dilution that was plated on the agar plate.

Answers to Simple Dilution Problems 1)
2) 3)

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answer key

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worksheet 2

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ChemTeam: Dilution
Problems #1-10

Dilutions Worksheet

1) If I add 25 mL of
water to 125 mL of a
0.15 M NaOH

solution, what will
the molarity of the
diluted solution be?

2) If I add water to
100 mL of a 0.15 M
NaOH solution until

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the final volume is 150 mL, what will the molarity of the diluted solution be?

3) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

Quiz & Worksheet -
How to Calculate
Dilution ... -

Study.com

1 Dilutions

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Worksheet Name:

Key KEY _____ 1.

When a solution of glucose, $C_6H_{12}O_6$, is diluted, the number of moles of the solute in the original solution is (greater than, less than, the same as) the number of moles of solute in the resulting less concentrated

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Problems Answer
Key

solution. 2.

Calculate the molarity of the resulting solution if a certain volume of water was added to 50.0 mL of 2.10 M

CHEMISTRY

DILUTION

PRACTICE - Miami-
Dade County Public
Schools

Molarity and

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Problems Answer

Dilutions Practice

Problems €

Molarity =

moles solute

Liters solution

Molarity 1

$x \text{ Volume} = \text{Molarity } 2$

$x \text{ Volume } M_1 V_1 = M_2$

$2 V_2$ 1) How many

grams of potassium

carbonate, K_2CO_3 ,

are needed to make

250 mL of a 2.5 M

solution? 1st

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Problems Answer

Key
calculate the moles
of solute 2nd use
moles of solute to
convert to grams of
solute 1) € $2.5M = x$
 $0.25L \times \dots$

Molarity & Dilutions Practice

ProblemsKEY

dilution from an 18.0
M H_2SO_4 stock
solution? Answers:

1. 39.2-ml (Put in

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Problems Answer

Key
paragraph form) 2.

1.1 M 3. 4.64 M 4.

49.2-ml Take 49.2-ml

of 18.0 M H_2SO_4

stock solution and

pour it into a 500-ml

volumetric flask. Fill

to the 500-ml line

with distilled water

to make 1.77M H_2SO_4

solution. Extra

Molarity Problems

for Practice 1.

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Dilution Worksheets

- Kiddy Math

This chemistry

video tutorial

explains how to

solve common

dilution problems

using a simple

formula using

concentration or

molarity with

volume. This video

also provides the

equations needed to

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Key

SOLUTIONS
WORKSHEET 2
MOLARITY AND
DILUTION
PROBLEMS ...
Dilutions and
Concentrations
Learning Objective.
1. Learn how to
dilute and
concentrate
solutions. Often, ...

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Problems Answer

Key Takeaways. ...

Answers. 1. Dilution is a decrease in a solution's concentration, ...

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