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1. protons; neutrons 2. electrons 3. neutrons 4. electrons 5. ionic 6. The two main types of chemical bonding are ionic and covalent bonding.

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Write your answer in the space provided. Date Class 11. List the five traditional branches of chemistry and explain what each involves. 12. Discuss the role of chemistry in one of these areas: Materials, Energy, Medicine and Biotechnology, Agriculture, the Environment, or the Universe. 13. looks unhealthy and that, perhaps, the grass is dying.

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Core Teaching Resources . Name Date Class IONIC AND METALLIC BONDING Practice Problems In your notebook, answer the following SECTION 7.1 IONS For each element below, state (i) the number of valence electrons in the atom, (ii) the electron dot structure, and (iii) the chemical symbol(s) for the most

05 CTR ch01 7/9/04 3:24 PM Page 7 CHEMISTRY 1

Choose the best answer and write its letter on the line. 11. The fourth principal energy level has a. 4 orbitals. b. 16 orbitals. c. 32 orbitals. . 9 orbi s. 12. If the electron configuration of an element is $1s^2 2s^2 2p^6 3s^2 3p^5$, the 116 element is a. iron. b. bromine. Core Teaching Resources c. chlorine. d. phosphorus.

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25.1 Nuclear Radiation 25

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