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You won't find fiction here  like Wikipedia. Wikibooks is devoted entirely to the sharing of knowledge.

Dimedone - Wikipedia  
Tributyltin hydride is an organotin compound with the formula (C 4 H 9) 3 SnH. It is a colorless liquid that is soluble in organic solvents. The compound is used as a source of hydrogen atoms in organic synthesis

Tributyltin hydride - Wikipedia  
Organic Chemistry, J. Clayden, N. Greeves, S. Warren, P. Wothers, Oxford, Oxford, 2001, cap. 17 e 19. Mecanismo de uma reação S N2: Substituição nucleofílica bimolecular (S N2) é um processo de uma etapa: o nucleófilo ataca o halo-alcano com simultânea expulsão do grupo de partida.

8. Introdução a Reações Orgânicas  
If your fingers are cold or your muscles ache, you can use chemical hand warmers to heat them up. There are two types of chemical hand warmer products, both using exothermic (heat-producing) chemical reactions. Here is how they work.

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Dimedone is a cyclic diketone used in organic chemistry to determine whether a compound contains an aldehyde group. Cyclohexanediones in general can be used as catalysts in the formation of transition-metal complexes. Other uses include applications in colorimetry, crystallography, luminescence and spectrophotometric analysis.

Dimedone - Wikipedia  
Dimedone is a cyclic diketone used in organic chemistry to determine whether a compound contains an aldehyde group. Cyclohexanediones in general can be used as catalysts in the formation of transition-metal complexes. Other uses include applications in colorimetry, crystallography, luminescence and spectrophotometric analysis.

How Chemical Hand Warmers Work - ThoughtCo.com  
Dimedone (orthoester) is a cyclic diketone used in organic chemistry to determine whether a compound contains an aldehyde group. Cyclohexanediones in general can be used as catalysts in the formation of transition-metal complexes. Other uses include applications in colorimetry, crystallography, luminescence and spectrophotometric analysis.

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