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Chemquest 31 Answer Key, Chemquest 30 Intro To Moles ... are the most involved in chemical reactions and bonding. The electron configuration for thallium (#81) is: $1s^2 2s^2 2p^6 3s^2 3p^6 4s^2 3d^{10} 4p^6 5s^2 4d^{10} 5p^6 6s^2 4f^{14} 5d^{10} 6p^1$. The outermost energy level (not sublevel) is the 6th energy level. How many total electrons does thallium have in the sixth level 3, they are in boldfaced ...

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1. a) How many moles of water are produced when 1.45 moles of propane
are combusted? b) How many molecules of water is this? (Remember each
mole has 6.02×10^{23} molecules.) 2. If 2.35 moles of CO_2 are produced
in a reaction, how many moles of H_2O would be produced? 3. Why is
this statement false: "If 10 grams of propane burn, you need 50 ...

ChemQuest 33
Does the reaction depend on the concentration of ClO_2 and the
concentration of OH^- - equally In other words, is the rate more
dependent on ClO_2 , on OH^- , or is it equally dependent on the
concentration of both.

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Answers To Chemquest 32 Moles And Reactions

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12 moles. For every mole of propane that combusts 3 moles of CO₂ are produced, so the number of moles of CO₂ that can be produced when 4 moles of propane combusts = 4(3) = 12. Verify that if 12.5 moles of propane and 63.2 moles of oxygen were present, then propane is the "limiting reactant" and oxygen is the excess reactant.

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Here is what a chemical reaction looks like when it is written
 $\text{Cu}(\text{NO}_3)_2 + \text{Zn} \rightarrow \text{Zn}(\text{NO}_3)_2 + \text{Cu}$ There is a sentence that you can write to describe what happens above: Copper(II) nitrate reacts with zinc metal and yields zinc nitrate and copper metal. Take a minute to notice the relationship between the sentence and the symbols in the reaction

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