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what its properties are and how it changes  
matter is anything that has mass and takes up  
space that is anything that

d A rock is broken into small pieces

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John R. Holum. just nineteen chapters in all,  
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a) Physical Change: In this type of changes, there are no permanent changes in the structure of matters. Physical appearance of the matters are changed. In physical changes, matters can return its old structure, no new matter is produced, and chemical structure of the matters does not change. For example, melting of ice, breaking glass.

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study of the composition, structure, and properties of matter and the changes it undergoes. Chemistry deals with questions such as what is something made of, why did something react? What is a material's internal make-up, and how does it behave when heated, cooled, or mixed with another substance?

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Matter undergoes phase changes or phase transitions from one state of matter to another. Below is a complete list of the names of these phase changes. The most

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commonly known phase changes are those six between solids, liquids, and gasses. However, plasma also is a state of matter, so a complete list requires all eight total phase changes.

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Chemistry = Matter and its change 17 Two important characteristics about matter and change q Mass is conserved. q Energy is required to affect change Law of conservation of mass 18 q Antoine Lavoisier q "Matter is neither created nor destroyed in a chemical

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reaction" q The total amount of matter present before a chemical reaction is always the same as the total amount after.

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matter what it consists of what its

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properties are and how it

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