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Ca(NO<sub>3</sub>)<sub>2</sub> Mg<sup>+2</sup> MgCl<sub>2</sub> MgCO<sub>3</sub> Mg(OH)<sub>2</sub> MgSO<sub>4</sub> Mg<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub>  
Mg(NO<sub>3</sub>)<sub>2</sub> Zn<sup>+2</sup> ZnCl<sub>2</sub> ZnCO<sub>3</sub> Zn(OH)<sub>2</sub> ZnSO<sub>4</sub> Zn<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> ...

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Writing Formulas (KEY) (Criss-Cross Method)

13. Answer questions 2.b., 2.c., and 2.d. of "Practice Problem" on  
page 44 of the textbook. Explain how you arrived at your answer.

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multivalent element: an element with more than one stable ion  
Polyatomic Ions Example: NaOH(aq) The cation is Na<sup>+</sup>, and the anion is OH<sup>-</sup>. The anion is made up of two

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