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Chapter 2 The Structure of Matter
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202 – 209 CHAPTER Chemistry:
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correct symbol listed below to
describe each of the chemical
reactions listed in questions 1–15.
Also write a balanced chemical
equation for each reaction. S =
synthesis SD = single
displacement

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Equations CHAPTER 6

14 Study Guide for An

Introduction to Chemistry

Exercise 2.3 - Cations and Anions:

Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a

magnesium atom with 12 protons and 10 electrons $(+12) + (-10) =$

$+2$ This is a $+2$ cation.

a fluorine atom with 9 protons and 10

electrons

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3.1 The Development of the
Periodic Table 3 Use with text
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just once to complete the
following passage.

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