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Date Class 6.1 Chemical Equations CHAPTER 6

14 Study Guide for An Introduction to Chemistry Exercise 2.3 - Cations and Anions: Identify each of the following as a cation or an anion, and determine the charge on each. (Obj 22) a magnesium atom with 12 protons and 10 electrons $(+12) + (-10) = +2$ This is a +2 cation. a fluorine atom with 9 protons and 10 electrons

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Name Date Class STUDY GUIDE 3.1 The Development of the Periodic Table 3 Use with text pages 86 - 94 CHAPTER Chemistry: Concepts and Applications Study Guide, Chapter 39 Use each of the terms below just once to complete the following passage.

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comprehension and application of concept; Thinking Critically ...

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CHAPTER Chemistry: Concepts and Applications Study Guide, Chapter 621
In each blank, write the word or phrase that best completes the following passage. When a piece of magnesium metal is added to dilute hydrochloric acid, fizzing occurs and hydrogen gas is released from the ...

Chapter 2 The Structure of Matter and the Chemical Elements

STUDY GUIDE 6.2 Types of Reactions 6 Use with text pages 202 - 209

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CHAPTER Chemistry: Concepts and Applications Study Guide, Chapter 623
Choose the correct symbol listed below to describe each of the chemical reactions listed in questions 1-15. Also write a balanced chemical equation for each reaction. S = synthesis SD = single displacement

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