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AP Chemistry > Chapter 5
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Flashcards in Chapter 5
Practice Test Deck (25) 1
Using the barometer shown in the figure, determine the pressure in terms of atmosphere. 0.576 atm 2
Which of the following statements about gases is not true?

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Chemistry (Chapter 5
Practice Test) Matching
Match each item with the
correct statement below. a.
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AP Chemistry: Practice Test,
Ch. 5. - Gases Name\_\_\_\_\_

MULTIPLE CHOICE. Choose the
one alternative that best
completes the statement or
answers the question.

1) According to kineticmolecular theory, in which
of the following gases will
the root-mean-square speed
of the molecules be the
highest at 200eC? A) SF6
B) H2O C) HCl D) Cl2

#### Chemistry Chapter 5 Practice Test -

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AP Chemistry: Practice Test, Ch. 5. - Gases MULTIPLE ... test for radioactivity. d. use of X rays. \_\_\_\_ 6. Mendeleev did not always list elements in his

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periodic table in order of increasing atomic mass because he grouped together elements with similar. a. properties. c. ... Chemistry Chapter 5 Exam ...

#### Chapter 5 Standardized Test Practice Answers

Science is a wonderful subject to learn about - whether it's for career purposes or you're merely curious about the ins and outs of our universe, including what makes up the substances that comprise who we are and what we see daily. Today we'll be looking at science from a chemical standpoint, seeing if you can identify basic

chemistry concepts and balance equations. How will you perform ...

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SCH3U Riverdale Collegiate Institute Name: Chapter 5/6 PRACTICE Test 1) Answer the following questions in the spaces provided on the extreme right of the page. You do not have to show your work. Significant figures and appropriate units are required. /6 a) 1.0 g of NaCl contains grams of sodium b) The mass of 2.4 x 10 25 atoms of sodium is: c) 5.0 g of CaCO 3 contains moles of ...

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Practice Test Author: lqhsyz
qw.odysseymobile.co-2020-1102T00:00:00+00:01 Subject:
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#### Chemistry Chapter 5 Practice Test - code.gymeyes.com

AP Chemistry Test (Chapter 5) Please do not write on this test. Please use the answer sheet. Multiple Choice 1) A sample of argon gas is sealed in a container. The volume is doubled, while the temperature is held constant. What has happened to the pressure on the gas? A) It is halved. B) It is reduced by 1/4 th. C) It is doubled.

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