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potassium perman-ganate ( $\text{KMnO}_4$ )  
0.01 M sodium hydrogen sulfite  
( $\text{NaHSO}_3$ ) Observing a Chemical  
Reaction Reactants are consumed in a  
chemical reaction as products are pro-  
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exist with all chemical reactions that allow scientists to predict amounts of products formed, reactants consumed, and percent yield based on theoretical maximum.

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- While you should practice working as fast as possible, it is more important at this point in the course,

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that you practice without a calculator, even if it slows you down. Look for the “ easy math ” - common factors and rough estimation - do not do “ long division ” to try to get exact values. Remember it is a MC test, use the answers

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calculation of quantities in chemical reactions is a subject of chemistry.

Mole ratio. A conversion factor derived from coefficients of a balanced chemical equation interpreted in terms of moles.

Limiting reagent . The reagent that determined the amount of product

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that can be formed by a ...

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