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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: CH₃CO₂H a. C₂H₆O 4 N₂O₅ b. N₂O₅ HgCl c. Hg₂Cl₂ d. C₆H₁₂ 2. C₄H₈ A certain hydrocarbon has an empirical formula of CH₂ and ...

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12. Name three characteristics of most metalloids. ... CHAPTER 1 REVIEW Matter and Change MIXED REVIEW ... For each type of investigation, select the most appropriate branch of chemistry from the following choices: organic chemistry, analytical chemistry, biochemistry, theoretical chemistry. More than one branch may be appropriate.

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CHAPTER 12 REVIEW Solutions MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Solid CaCl₂ does not conduct electricity. Explain why it is considered to be an electrolyte. CaCl₂ is an ionic solid. In the crystal form, its ions are locked in position. Dissolving

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12. A bond's character is more than 50% ionic if the electronegativity difference between the two ... CHAPTER 6 REVIEW Chemical Bonding SECTION 3 ... MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

7 Chemical Formulas and Chemical Compounds

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Modern Chemistry 1 Solutions CHAPTER 12 REVIEW Solutions Teacher Notes and Answers Chapter 12 SECTION 1 SHORT ANSWER 1. c 2. a 3. b 2. a. alcohol b. water c. the gels 3. The mixture is a colloid. The properties are consistent with those reported in Table 3 on page 404 of the text. The particle size is small, but not too small, and the mixture

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1 Matter and Change - HUBBARD'S HONORS CHEMISTRY CLASS

CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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Chemistry Chapter 12 Mixed Review

Chapter 12 Review, Mixed Review: Solutions Although there are only six questions on this chemistry handout, it makes a thorough review of solutions. Novices explain why a compound is not an electrolyte, identify types of compounds, and calculate moles, grams, and molalities in various solutions.

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