

# Chemistry A Bonding Packet Answers

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### Answer Key To A Chemical Bonding Packet

Chemical Bonding and Molecular Structure 2 Level 2--Orbital Theory – electrons and + nucleus of one atom strongly perturb or change the spatial distribution of the other atom's valence electrons. A new orbital (wave function) is needed to describe the distribution of the bonding electrons bond orbital  $\sigma$  bond orbital--describes the motion of the 2 electrons of opposite spin

### AP\* Chemistry CHEMICAL BONDING & MOLECULAR STRUCTURE

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Chemistry A Bonding

3. The bonding found inside a molecule of carbon tetrachloride is ... 4. The bonding that holds water molecules together to make ice is ... 5. The bonding found in a white, high melting point crystalline solid that conducts electricity when liquid... 6. The bonding found in diamonds is ... 7. The bonding found between atoms in carbon disulfide ...

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The bonding between the carbon atoms is The attraction of a hydrogen atom in one molecule for a more electronegative element in another molecule is what we call The bonding that is synonymous with dispersion forces is The strongest of the above choices is The weakest of the above choices is The bonding that involves two atoms sharing three pairs of electrons is The bonding that is broken when ...

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Ionic Bond: Na-F bond. Na is a metal and has a low electronegativity (attraction of electrons) while F has a high electronegativity. Therefore, fluorine will take an electron from sodium in an ionic bond. Now, try it on your own. For the following compounds or molecules, determine if they are covalent or ionic.

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