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Student Laboratory Kit Introduction Just

as a fingerprint is unique to each person,

the color of light emitted by metals heated

in a flame is unique to each metal.

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If the vinegar used in the lab had 5.89 g of acetic acid per 100. mL of vinegar, what volume (in mL) of 0.0998 M NaOH would you use to reach the endpoint of the titration? ... Get a free answer to a quick problem. Most questions answered within

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relative loudness of pure oxygen and pure

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hydrogen in the pop-test. 3. Write a balanced chemical equation for the combustion reaction of hydrogen and oxygen to give water.

Chemistry

AP Chemistry Laboratory #21

Introduction Oxidation-reduction

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reactions form a major class of chemical reactions. From the reactions of oxygen with sugars, fats, and proteins that provide energy for life to the corrosion of metals, many important reactions involve the processes of oxidation and reduction.

Analysis of Alum Lab Explanation

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of this laboratory is to construct a table listing the reduction potentials of a series of metal ions, in order of ease of reduction. The series of microscale half-cells is constructed by placing a piece of metal into a 1.0 M solution of its ions for each metal in the series. The metals chosen are copper, iron, lead, magnesium,

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silver, and zinc.

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equilibrium plays an important role in
acid-base chemistry and in solubility.

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(Enduring Understanding 6C) 6C1:
Chemical equilibrium reasoning can be used to describe the proton-transfer reactions of acid-base chemistry.

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wins two R&D 100 Awards (Posted
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AP Chemistry Laboratory #21

Pre-Lab Questions (Answer questions on a separate sheet of paper.) 1. Combine the oxidation and reduction half-reactions for hydrogen peroxide and permanganate ion, respectively, and write the balanced chemical equation for the overall reaction

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between H_2O_2 and MnO_4^- – in acid solution. Hint: The number of electrons transferred must ...

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Introduction viii PS-2877PS-2877 inquiry
possibilities for students' investigations
see the suggestions in Using these Labs

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with the AP and the IBO Programs in this Introduction. Additionally, this manual presents teacher-developed laboratory activities using 21 st-century technologies to help you and your students explore topics, develop scientific inquiry skills, and

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Chromatography Challenge continued 2

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General Procedure 1. Obtain a piece of filter paper. Using a sharp pencil or a pushpin, poke a small hole in the center of the filter paper.

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Chromatography Challenge

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determine the specific types of pigments

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