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Chapter 8

Solutions Acids

Chapter 8
Bases Wordwise

Solutions

Acids Bases

Wordwise

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Page 8/39

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Solutions, Acids

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Chapter 8

Solutions,

Acids, and Bases

Summary 8.1

Formation of

Solutions

- A solute is a substance whose particles are dissolved in a solution.
- The substance in which the solute dissolves is

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called the
solvent.

Substances can
dissolve in
water in three
ways—by
dissociation,

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Acids, and Bases
- Practice Test
...

Chapter 8 Acids,
Page 10/39

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Chapter 8

Solutions Acids
Bases Wordwise
Bases, and Acid-
Base Reactions

... Chapter Map.

Arrhenius Base

Definitions • A

base is a

substance that

generates OH^- ...

solution is

neutral in the

acid/base sense.

pH • To avoid the

small numbers

associated with

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describing
Bases Wordwise
acidic and basic

solutions in
terms of

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Answers

*Unit 12: Chapter
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Page 13/39

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Answer Key.

Define: Solute – a substance whose particles are dissolved in a solution.

Solvent – a substance in which a solute dissolves.

Solution – a mixture that forms when substances

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*dissolve and
form a*

*homogeneous
mixture.*

*Identify and
explain what
each part is for
lemonade.*

*solutions
chapter 8 acids
bases 8*

*Flashcards and
Study ...*

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- Chemistry

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Section 8.1:

Formation of

Solutions I.

Dissolving Group

Page 17/39

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#_____ Main

Idea:

Drawing_____ A.

Dissociation of
Ionic Compounds
Group ...

Quia - Chapter
8, Solutions,
Acids, and Bases
Ch 8 Solutions,
Acids and Bases
Study Guide 16.
For a solution

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Solutions Acids Bases Wordwise

to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must _____. 17.

During the formation of a solution, energy is _____. 18.

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Base Reactions
Review Skills
8.1 Strong and
Weak Acids and
Bases Acid*

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Review Bases

*Internet: Strong
and Weak Bases*

Internet:

*Identification
of Strong and
Weak Acids and
Bases Special*

Topic 8.1:

*Chemistry and
Your Sense of*

*Taste 8.2 pH and
Acidic and Basic
Solutions*

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8--Solutions,

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different sets
of solutions
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bases 8
flashcards on
Quizlet.

Ch 8 Solutions,
Acids and Bases
Study Guide
Solutions that
are neither
basic nor acidic
are neutral.

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Bases Wordwise

This chapter discusses the nomenclature of common acids and identify some important bases. Pure acids and bases and their concentrated aqueous solutions are commonly encountered in the laboratory.

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Bases Wordwise

They are usually
highly
corrosive, so
they must be
handled with
care.

Chapter 8 Acids,
Bases, and pH -
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Solutions,

Page 27/39

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Chapter 8

Solutions Acids
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Acids, and Bases
Section 8.2

*Solubility and
Concentration*

(pages 235–239)

*This section
explains*

*solubility, the
factors*

affecting

*solubility, and
different ways*

*of expressing
the*

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Bases Wordwise
concentration of
a solution.

*Chapter 8 Acids,
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*Chapter 8 Acids,
Bases, and pH
Solutions for
Practice*

Problems Student

Textbook page

382 1. Problem

Name and write

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Bases Wordwise

*the formula of
the conjugate
base of each
molecule or ion.*

*Chapter 8:
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Acids, and Bases
Test and improve
your knowledge
of Chapter 8:
Solutions,
Acids, and Bases
with fun*

Page 30/39

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vocabulary;

Prentice Hall;

Chapter 8.

STUDY. PLAY.

solute.

*substance that
is dissolved in
a solvent to
make a solution.*

solvent.

*substance in
which a solute
is dissolved to
form a solution.*

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Solutions Acids

dissociation.

Bases Wordwise

*Chapter 8 Acids,
Bases, and Acid-
Base Reactions*

*The numbers of
moles of solute
dissolved in a
liter of*

solution. acid:

*A solution that
contains*

hydrogen ions.

indicator: Any

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Solutions Acids Bases Wordwise

substance that changes colors in the presents of acids or bases. base: A solution that forms OH.

neutralization: A chemical reaction between an acid and a base. salt: An ionic compound formed when

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Bases Wordwise

an acid reacts
with a ...

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Acids, and Base

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Page 35/39

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*helps students
learn the
essential
lessons
associated with
acids, bases and
solutions.*

*Section 8.2
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Lincoln*

*Interactive
Ch 8 Solutions,
Page 36/39*

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Answers 16. For a solution to form, one substance must dissolve in another. For this to happen, the solute and solvent particles must ATTRACT ONE ANOTHER. 17.

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During the formation of a solution, energy is ABSORBED & RELEASED. 18.

The solubility of solids increases as the temperature of the solvent ...

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Page 38/39

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