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CHAPTER 7

HOMEWORK 7-2 ...

CHAPTER 7 •

HOMEWORK 7-2

VOCABULARY 1. d 2.

c 3. e 4. a 5. g 6. h 7.

b 8. f GRAPHIC

ORGANIZER Name

of Compound or Ion

Formula of

Compound Copper(II)

ion Cu^{2+} Ammonium

ion NH_4^+

Bismuth(III) bromide

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BiBr_3 Stannous
fluoride SnF_2

Magnesium fluoride

MgF_2 Iron(III)

cyanide $\text{Fe}(\text{CN})_3$

PPT – Modern

Chemistry Chapter 7

Chemical Formulas ...

CHAPTER 5 REVIEW

The Periodic Law

SECTION 1 SHORT

ANSWER Answer the

following questions in

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the space provided. 1.
c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered.

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Chemical Bonding
MIXED REVIEW
SHORT ANSWER
Answer the following
questions in the
space provided. 1. a.

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7 Chemical Formulas
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Chemical Bonding

CHAPTER 7 REVIEW

Chemical Formulas
and Chemical

Compounds

SECTION 3 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

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Label each of the following statements as True or False:

_____ a. If the formula mass of one molecule is x u, the molar mass is x g/mol. _____ b.

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Chemistry Chapter 7
Chemical Formulas 1

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Chapter 7 Chemical Formulas Chemical Compounds . A chemical formula indicates the kind and relative number of atoms in a chemical compound. C_8H_{18} (octane) has 8 carbon and 18 hydrogen atoms. 2 Forming Ionic Compounds. Compounds that have

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the elements held
together by

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you are still
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want to cheat.

5 The Periodic Law
Chemistry Chapter
Resources. Chapter
One Matter &
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CHAPTER 7 REVIEW

Chemical Formulas

Online Library
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Modern Chemistry
and Chemical
Answers
Compounds

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron(III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be

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attached to the iron
atom. (c) the charge
on ...

7 Chemical Formulas
and Chemical
Compounds

CHAPTER 7 REVIEW

Chemical Formulas
and Chemical

Compounds MIXED

REVIEW SHORT

ANSWER Answer the
following questions in

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Answers

the space provided. 1.

Write formulas for the following compounds:

CuCO₃ a. copper(II)

carbonate Na₂SO₃

b. sodium sulfite (NH₄)₃PO₄ c.

ammonium phosphate

SnS₂ d. tin(IV)

sulfide HNO₂ e.

nitrous acid 2.

6 Chemical Bonding

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